4061- Lecture Three

Two Level Atoms in Thermal Equilibrium

- Populations of levels remain constant
- A balance rate of upwards and downward transitions



$$n_2 = n_1 e^{E_{21}/k_B T} = n_1 e^{hv_0/k_B T}$$
 (1)

Blackbody or Plank Spectrum



 \circ constant over atomic Lineshape g(v)

$$\frac{dn_2}{dt} = n_1 B_{12} \rho_v - n_2 B_{21} \rho_v - n_2 A_{12}$$
(3)

- terms $B_{12} \rho_{v}$, $B_{21} \rho_{v}$, A_{21} have dimensions of Rate [s⁻¹]
- For a steady state/equilibrium $dn_2/dt = 0$ (4)
- Using (4), substitute (1) into (3), and solve for n_2/n_1

$$e^{-\frac{h\nu_{o}}{k_{B}T}} = \frac{B_{12}\rho_{\nu}}{A_{21}} + B_{21}\rho_{\nu} (5)$$

- This is true for all temperatures
- A and B coefficients are independent of temperature as they are properties of the atom

For High Temperature

- Rayleigh Jean's Law $\rightarrow \rho_{v} \sim k_{\rm B}T$
- For High temperature $k_BT \gg h\nu$ which implies $\rho_{\nu} = large$
- In equation 5, this implies that $B_{21} \rho_{\nu} >> A_{21}$
- Therefore (5) becomes:

$$e^{-0} = 1 = B_{12} \rho_{\nu} / B_{21} \rho_{\nu}$$
$$\Rightarrow B_{12} = B_{21}$$

- $B_{21} = B_{12}$, solve for ρ_v in equation 5

$$\rho_{\nu} = \frac{A_{21}}{B_{21}} \left(\frac{1}{\frac{h\nu_0}{e^{k_B T} - 1}} \right) (6)$$

- Comparing this with Planks distribution (equation 2) a relationship between the spontaneous emission rate and the stimulated emission rate can be found

$$A_{21}/B_{21} = 8\pi h v_o^3 / c^3$$

- Higher A coefficient implies a higher B coefficient. So a transition with a strong decay rate can be a good laser transition

Interpretation

$$\rho_v = \text{Energy}/(\text{Vol})(\Delta v) = [\text{Modes}/(\text{Vol})(\Delta v)] [\text{Energy}/\text{Photon}] [\text{Photons}/\text{Mode}]$$

$$\rho_{v} = [\beta_{v}] [hv_{o}] [\overline{N}]$$

$$\rho_{v} = (8\pi v_{o}^{2}/c^{3})(hv_{o}) \left[\frac{1}{\frac{hv_{o}}{k_{B}T}} - 1\right]$$

 $- 1/(e^{hv_o/k_BT}-1)$ is the occupation number for bosons (photons)

- For fermions it is +1 in the denominator rather than -1

Stimulated Emission Rate

$$\begin{aligned} R_{21} &= \rho_{\nu} \ B_{21} \ (7) \\ \beta_{\nu} \ h\nu &= \ 8\pi h\nu^3/c^3 = A_{21} \ / \ B_{21} \end{aligned}$$

Solving for B_{21} and plugging into (7), it can be shown that