## Phys 4061/5061 – Tutorial One

Topics Covered in Tutorial

- Properties of Diode Lasers
- Principles of Faraday Isolators
- Overview of Rubidium Spectroscopy

Details Pertaining to laboratory experiments covered in this tutorial can be found in the lab manual under the following sections

- 1. Absorption/Emission Spectroscopy / EOM
- **2.** Farby-Perot Laser Linewidth
- 3. Lockin Amplifier
- **4.** Zeeman Shift

Lasers used in Laboratory experiments are external cavity diode lasers (ECDLs) ECDLs used in laser diodes similar to those in CD players.

Review ECDL components and their functions

- 1. Cavity
- 2. Controllers for:
  - Scan
  - Temperature
  - Current
- 3. Role of grating in providing optical feedback to narrow line width
  - Through the process of stimulated emission the diode is forced to emit laser light over a narrow frequency range selected by the grating

Review Properties and Applications of a Typical External Cavity Diode Laser

- 1 MHz Linewidth
- 0.2 mA Current Drift
- 1/500 °C Temperature Drift
- 100 mW Power
- 1 % Intensity Variation

Laser Box



- need stable current to control intensity and frequency of light
- Need stable temperature and scan control to co troll the length of external cavity
- That determines frequencies of cavity modes
- A faraday isolator protects EDCLs from optical feedback
- Labs that use EDCLs include
  - 1. Lock-in Amplifier
  - 2. Zeeman Shift/Power broadening
  - 3. Fabry Perot Laser Linewidth
  - 4. absorption/emission spectroscopy/EOM

## Why Use Rubidium?

- Principle resonance line matches the energy band-gap in diode laser at 780nm

Rb vapour at room temperature is a good approximation to an ideal gas. The optical depth of a 5cm vapour cell is of order unity. So the atomic response can be measured with adequate signal to noise in introductory undergraduate experiments on spectroscopy.

- A rubidium cell at room temperature contains 72% <sup>85</sup>Rb and 28% <sup>87</sup>Rb
- Rb resides in Group IA of the periodic table and represents a 1 electron atom
- It is reactive toxic and easily oxidized
- Vapour pressure at STP ~  $3 \times 10^{-7}$  T